Acta Crystallographica Section E Structure Reports Online ISSN 1600-5368 Editors: W. Clegg and D. G. Watson

# *N*,*N*-Diisobutyl-*N*'-(2-thienylcarbonyl)thiourea

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Acta Cryst. (2007). E63, o1158-o1159

Gomes *et al.*  $\cdot$  C<sub>14</sub>H<sub>22</sub>N<sub>2</sub>OS<sub>2</sub>

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#### **Key indicators**

Single-crystal X-ray study T = 290 KMean  $\sigma(C-C) = 0.004 \text{ Å}$ Disorder in main residue R factor = 0.041 wR factor = 0.107 Data-to-parameter ratio = 15.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## N,N-Diisobutyl-N'-(2-thienylcarbonyl)thiourea

The title compound,  $C_{14}H_{22}N_2OS_2$ , contains a disordered isobutyl group. There are no direction-specific interactions between the molecules of the title compound.

Received 24 January 2007 Accepted 30 January 2007

#### Comment

Substituted *N*-acylthioureas have been a subject of investigations, due to their ability to form stable metal complexes, their extraction proprieties and as model compounds in physical chemistry studies (Beyer *et al.*, 1981; Mühl *et al.*, 1986). We report here the structure of *N*,*N*-diisobutyl-*N'*-(2-thienylcarbonyl)thiourea, (I) (Figs. 1 and 2). The main bond lengths are given in Table 1 and are within the ranges obtained for similar compounds (Bailey *et al.*, 1988; Koch *et al.*, 1995; Morales *et al.*, 1997).



The dihedral angle between the O2/C21/N1 and S1/C6/N2 planes is 39.5 (3)°, while that between the O2/C21/N1 plane and the thiophene ring is  $6.8 (4)^{\circ}$ 

#### **Experimental**

The title compound, (I), was prepared by adapting the general procedure described by Douglass & Dains (1934) for other *N*-acyl-



#### Figure 1

© 2007 International Union of Crystallography All rights reserved The structure of the major component of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

doi:10.1107/S1600536807004850

### organic papers

thioureas. A solution of anhydrous acetone containing thienyl chloride was added to an equimolar acetone–potassium thiocyanate solution. After stirring and boiling the mixture, a solution of equimolar diisobutylamine in acetone was added and kept under reflux for 1 h. The final mixture was poured into a water–ice bath, and crystals were formed. These were filtered and purified by recrystallization from an ethanol/water mixture. The resulting solid product (I) was crystallized from a dichlorometane–methanol (1:1) mixture yielding X-ray quality single crystals (m.p. 413.2 K). Elemental analysis for  $C_{14}H_{22}N_2OS_2$  found: C 55.7, H 7.5, N 9.6, S 22.1%; calculated: C 56.34, H 7.43, N 9.4, S 21.5%.

Z = 4

 $D_x = 1.215 \text{ Mg m}^{-3}$ 

 $0.3 \times 0.2 \times 0.1 \text{ mm}$ 

1 standard reflection

frequency: 80 min

intensity decay: none

 $w = 1/[\sigma^2(F_o^2) + (0.0359P)^2]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

+ 0.8882P]

 $\Delta \rho_{\rm min} = -0.31 \text{ e } \text{\AA}^{-3}$ 

 $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.26 \text{ e } \text{\AA}^{-3}$ 

2861 independent reflections

2161 reflections with  $I > 2\sigma(I)$ 

Mo  $K\alpha$  radiation

 $\mu = 0.32 \text{ mm}^{-1}$ 

T = 290 (2) K

Prism, yellow

 $\begin{aligned} R_{\rm int} &= 0.038\\ \theta_{\rm max} &= 25.0^\circ \end{aligned}$ 

#### Crystal data

 $\begin{array}{l} C_{14}H_{22}N_2OS_2\\ M_r = 298.4\\ \text{Monoclinic, } P2_1/c\\ a = 12.0720 \ (6) \ \text{\AA}\\ b = 8.5169 \ (7) \ \text{\AA}\\ c = 16.7544 \ (18) \ \text{\AA}\\ \beta = 109.243 \ (6)^\circ\\ V = 1626.4 \ (2) \ \text{\AA}^3 \end{array}$ 

#### Data collection

Stoe Stadi-4 four-circle diffractometer Profile-fitted  $2\theta/\omega$  scans (Clegg, 1981) Absorption correction:  $\psi$  scan (North *et al.*, 1968)  $T_{\min} = 0.888, T_{\max} = 0.968$ 5714 measured reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.041$   $wR(F^2) = 0.107$  S = 1.082861 reflections 186 parameters H-atom parameters constrained

#### Table 1

Selected bond lengths (Å).

S2-C6	1.677 (2)	N1-C21	1.383 (3)
S1-C5	1.700 (3)	N1-C6	1.421 (3)
S1-C2	1.724 (2)	N2-C6	1.328 (3)
O2-C21	1.218 (3)		

All the H atoms bonded to C atoms were assigned standard C–H distances [0.93 (aromatic), 0.98 (aliphatic)), 0.96 (methyl) or 0.97 Å (CH<sub>2</sub>), with  $U_{iso}(H) = kU_{eq}(C)$ , where k = 1.5 for methyl groups and 1.2 for other H atoms bonded to C atoms, and N–H = 0.83 Å, with  $U_{iso}(H) = 1.2U_{eq}(N)$ ]. The central atom of the isobutyl group containing atoms C7, C9 and C10 is disorded over two sites, *viz.* C8A and C8B. H atoms were calculated for atoms C7, C8, C9 and C10 on the basis of this disorder. Those for the methyl atoms C9 and C10 were calculated as six equally spaced half-occupancy hydrogens. The refined values of site-occupancy factors for the disordered





The structure of the minor component of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

groups attached to C8A and C8B are 0.642 (14) and 0.458 (14), respectively.

Data collection: *IPDS Software* (Stoe & Cie, 1996); cell refinement: *CELL* in *IPDS Software*; data reduction: *INTEGRATE* in *IPDS Software*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON*, (Spek, 2003); software used to prepare material for publication: *SHELXL93* (Sheldrick, 1993).

The authors thank the Institut fur Anorganische Chemie, Universitat Halle-Wittenberg, for allowing X-ray data collection. Bernd Schröder thanks Fundação Para a Ciência e Técnologia, Lisboa, Portugal, and the European Social Fund (ESF) under the 3rd Community Support Framework (CSF), for the award of a Post-Doctoral scholarship (PRAXIS XXI/ BPD 35131/2000). The authors acknowledge financial support from Fundação Para a Ciência e Técnologia (POCI/QUI/ 61873/2004).

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